

ChE 374—Lecture 2—Fluid Properties

Q: What is a fluid?

- Shear stress
- Gas properties
- Liquid properties
- Continuum assumption
- Basic Fluid Properties

- Density

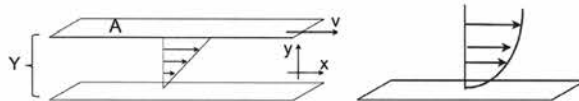
- * Water 1000 kg/m³, 62.3 lbm/ft³
- * Specific gravity
- * Specific weight
- * Industries: Degree API, Degree Baume
- * Ideal Gas
- * Liquid: Water: T=0-100°C → 4% variation! P=1-210 atm → 1% variation!

- Variation of ρ with P, T :

- * Pressure: $\kappa =$ Coefficient of compressibility: $\kappa = \rho \left(\frac{\partial \rho}{\partial P} \right)_T \rightarrow \frac{\Delta P}{\Delta \rho / \rho} \rightarrow \frac{\Delta \rho}{\rho} = \frac{\Delta P}{\kappa}$.
· I.G.?
- * Temperature: $\beta =$ Coefficient of volume expansion: $\beta = -\frac{1}{\rho} \left(\frac{\partial \rho}{\partial T} \right)_P \rightarrow -\frac{\Delta \rho / \rho}{\Delta T} \rightarrow \frac{\Delta \rho}{\rho} = \beta \Delta T$.
- * $\frac{\Delta \rho}{\rho} = \frac{1}{\kappa} \Delta P - \beta \Delta T$.

- Viscosity

Q: How to quantify it?



- * Shear Stress: $\tau = \mu \frac{dv}{dy}$
 - Sign convention F
 - Momentum Flux: $\tau = F/A$: Force is a Rate of Momentum!
 $F = ma = m \frac{dv}{dt} = \frac{dmv}{dt} = \frac{d(mom)}{dt}$.
 - Stress at wall → force at wall → mom. flux at wall → SLOPE AT WALL.
 - * Non-Newtonian fluids
 - τ vs dv/dx is nonlinear.
 - μ depends on strain rate
 - (Newtonian)
 - Bingham plastic
 - Pseudo-plastic
 - Dilatant
 - * Temperature effect
 - Liquids
 - Gases
 - * Pressure effect small for liquids and gases
- Kinematic viscosity: $\nu = \frac{\mu}{\rho}$ (=) m²/s.
- * μ (=) kg/m·s.

Class 2 - Fluid Properties

Q: What is a fluid?

- Gas or liquid
- Deforms continuously under applied shear stress

→ What is a shear stress (more later)

- Compressive → ←
- Tensile ← →
- Shear → → (Force parallel to surface)
- Stress = F/A (Like pressure)

- Glass?
- Asphalt?

Liquids: Form a free surface

Gases: Fill Volume, no free surface, mix w/ other gases.

Continuum:

- Fluids made of distinct molecules.
- Study fluids → normally ignore molecules. Homogenous.
- Continuum is valid when scales of interest \gg space between molecules, or time between collisions.
- mean free path of O_2 at 1 atm → 10^{-8} m
- Invalid at low P, or high speed, or small channels.

Basic Properties

- Name / List some, then discuss

- Density

- ρ
- kg/m^3

- Memorize water/air

- 1000 kg/m^3
- 62.3 lbm/ft^3
- 1.2 kg/m^3
- 0.0752 lbm/ft^3

- Specific Gravity

$$SG = \rho / \rho_{H_2O}$$

- ρ_{H_2O} @ $4^\circ C = 1000$
- need T for careful work.

Specific Weight.

$$\gamma = \rho g \quad : \quad \text{N/m}^3$$

weight per unit volume

$$\frac{F}{V} = \frac{m \cdot a}{V} = \frac{m g}{V} = \rho g$$

Industries

- Degree API : American Petroleum Inst.

$$\text{Deg API} = \frac{141.5}{SG} - 131.5 \quad \rightarrow \quad \frac{141.5 \rho_{H_2O}}{\rho} - 131.5$$

Q: Why? $\rightarrow \frac{1}{\rho} \rightarrow$ lighter oils higher Deg API
 \rightarrow float above, more valuable.

- Deg. Baumé
- Brix Gravity
- etc.

Ideal Gas.

$$\rho = \frac{MP}{RT}$$

Mech. Eng. use $R \rightarrow \frac{R}{M}$

$$\rho \propto P, \quad \rho \propto \frac{1}{T}$$

Liquids

$\rho \approx$ constant w/RT P , not T

4% variation $T = 0 - 100^\circ\text{C}$

1% variation $P = 1 - 210 \text{ atm}$

See Slides! K - coef. of compressibility.

$$K = \rho \left(\frac{\partial \rho}{\partial P} \right)_T \approx \frac{\Delta \rho}{\Delta P / \rho} \rightarrow \frac{\Delta \rho}{\rho} = \frac{1}{K} \Delta P$$

$K = f(P)$

$K \rightarrow$ measures fractional change in Density (Volume) w/P

$$\text{also } \frac{\Delta V}{V} = - \frac{\Delta P}{K}$$

$K = P$ for ideal Gas.

β : Coef of Volume Expansion

$$\beta = \frac{1}{V} \left(\frac{\partial V}{\partial T} \right)_P \rightarrow - \frac{\Delta \rho / \rho}{\Delta T} \rightarrow \frac{\Delta V}{V} = \beta \Delta T$$

measures fractional change in volume (Density) with T

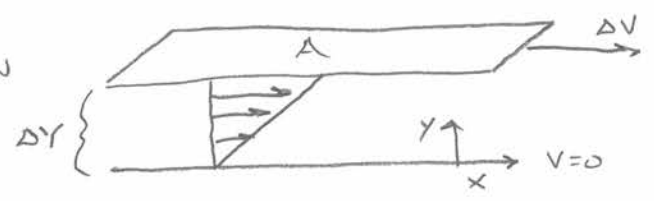
Viscosity

Q: what is visc? Honey vs water. **Movie**

Q: How to Quantity it.

*

Class activity: Draw



Force needed to keep Plate moving

- in terms of ΔV , A , ΔY

- Talk w/ neighbor \rightarrow Discuss

$F \propto A$, $F \propto \Delta V$, $F \propto \frac{1}{\Delta Y}$, μ is the \propto const.

$$\rightarrow F = \mu A \frac{\Delta V}{\Delta Y} \rightarrow \frac{F}{A} = \tau = \mu \frac{\Delta V}{\Delta Y}$$

For curved profile $\lim \Delta Y \rightarrow 0 \rightarrow \tau = \mu \frac{dV}{dy}$

Force of Fluid on Plate = - F of Plate on Fluid

$$\rightarrow \tau = -\mu \frac{dV}{dy} \quad \text{Memorize, know its parts.}$$

• μ is resistance to flow

• Newtonian Fluid. $\rightarrow \mu = \mu(T, P, \text{Fluid}) \quad \mu \neq \mu(v)$

• τ = Momentum flux: $\frac{F}{A} = \frac{\text{Rate of momentum}}{\text{Area}}$

Flux is Stuff Per time per area.

$$F = ma = m \frac{dV}{dt} = \frac{dmV}{dt} = \text{Rate of momentum}$$

T - effect on μ

Liquids : molecules always in the way
inc. T, molecules speed up \rightarrow motion easier

Gases : $\rightarrow \mu$ Dec.
molecules "never" in each others way, but
inc T \rightarrow in way $\rightarrow \mu$ inc.

P - effect on μ

Small for gases \rightarrow Liquids.

$\frac{\mu}{\rho} = \nu$ is kinematic viscosity $(=) \frac{m^2}{s}$

μ air vs water Factor 100

ρ air vs water Factor 1000

Non-Newtonian Fluids

